# Efficient stereoselective route to $\boldsymbol{\beta}$-lactams and their application to the stereoselective synthesis of a key intermediate for carbapenem antibiotic (+)-PS-5 ${ }^{1}$ 

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A combination of a stereoselective addition of benzenethiol to $\boldsymbol{\alpha}, \boldsymbol{\beta}$-unsaturated carboxylic acid derivatives and a subsequent substitution reaction of the corresponding sulfonium group with $O$-alkylhydroxamate anion has provided a new practical and stereoselective method for the construction of 3,4-disubstituted cis- and trans- $\beta$-lactams. A successful application was demonstrated by the formal asymmetric synthesis of (+)-PS-5.

## Introduction

The Michael-type addition of nucleophiles to $\alpha, \beta$-unsaturated carbonyl compounds is one of the most widely used reactions known. Recently we found that nucleophilic addition of thiols to electron-deficient olefins proceeds stereospecifically and with high stereoselectivity in the presence of excess thiol as a proton source, thus forming two contiguous stereogenic centers with a phenylsulfanyl substituent which has high potential for further conversion to various types of structures ${ }^{2}$ (Scheme 1). The


Scheme 1
addition is thought to proceed via the course of rapid protonation of the intermediary enolate formed by attack of the nucleophile (RSM). Thus the erythro-adduct was obtained stereoselectively from the $(E)$-olefin whilst the threo-adduct was obtained predominantly from the $(Z)$-isomer under the conditions employing an excessive amount of protic thiol. Furthermore, we have shown asymmetric total synthesis of $(+)$-diltiazem, ${ }^{3}(+)$-whisky lactone, ${ }^{4}(+)$-PS- $5^{1}$ and L-erythro-C18-sphingosine ${ }^{5}$ on the basis of the successful expansion of this addition reaction to an asymmetric addition reaction. ${ }^{2}$ In this paper, we describe full details of the stereoselective synthesis of substituted $\beta$-lactams and the asymmetric synthesis of a key intermediate for the synthesis of carbapenem antibiotic (+)-PS-5. ${ }^{6}$ Our approach is shown in Scheme 2. The synthetic strategy consists of two key steps: (i) stereospecific nucleophilic


Scheme 2
addition of the thiols $(\mathbf{1} \longrightarrow \mathbf{3}$ or $\mathbf{2} \longrightarrow \mathbf{4})$ and (ii) stereoselective displacement of the corresponding sulfonium group $\mathbf{( 3 \longrightarrow 5}$ or $\mathbf{4} \longrightarrow \mathbf{6}$ ). $S$-Alkylation of the erythro-adduct $\mathbf{3}$, generated from $(E)$-amide 1, and subsequent intramolecular and stereoselective displacement of the resulting sulfonium
group with $O$-alkyl hydroxamate anion $\mathbf{A}$ would furnish the trans- $\beta$-lactam 5. Since the $\mathrm{p} K_{\mathrm{a}}\left(\mathrm{p} K_{\mathrm{a}}: 6-10\right)$ of the $\mathrm{N}-\mathrm{H}$ in the hydroxamate moiety is lower ${ }^{7}$ than that ( $\mathrm{p} K_{\mathrm{a}}: c a$. 25) of the $\mathrm{N}-\mathrm{H}$ in the amide group, the desired cyclization of the $O$-alkyl hydroxamate anion is expected to proceed smoothly in the presence of a weak base without both competitive $\beta$-elimination and epimerization to provide trans- $\beta$-lactam 5. ${ }^{8}$ In a similar manner, the threo-adduct 4, prepared from $(Z)$-amide $\mathbf{2}$, would be converted into cis- $\beta$-lactam 6. Furthermore, the usefulness of this method can be demonstrated by the asymmetric synthesis of a key intermediate, ( $-\mathbf{- 9}$, for the synthesis of the carbapenem antibiotic (+)-PS-5 by using oxazolidinone as a chiral auxiliary.

## Results and discussion

## Stereoselective synthesis of cis- and trans- $\beta$-lactams

In the synthesis of carbapenem antibiotics, the difficulty in achieving control of the relative and absolute stereochemistries of the contiguous chiral centers and the construction of the $\beta$-lactam ring has remained unresolved. Particularly, the stereoselective and practical method for the construction of the cis-$\beta$-lactams has been unexploited compared with the many known ${ }^{8}$ synthetic works on the trans-lactams. We now provide a potential method for the stereoselective synthesis of both cis- and trans- $\beta$-lactams.

We first investigated conversion of methyl tiglate [ $(E)$-methyl 2-methylbut-2-enoate] into the trans-lactam as a model study (Scheme 3). Methyl tiglate $\mathbf{1 0}$ was treated with 10 equiv. of


Scheme 3 Reagents: i, 0.1 equiv. PhSLi, 10 equiv. PhSH; ii, BnO$\mathrm{NH}_{2} \cdot \mathrm{HCl}, \mathrm{Me}_{3} \mathrm{Al}$; iii, 1) MeI, $\left.\mathrm{AgClO}_{4}, 2\right) \mathrm{K}_{2} \mathrm{CO}_{3}$
benzenethiol in the presence of 0.1 equiv. of lithium benzenethiolate to give an inseparable mixture of the erythro- and threo-adducts $\mathbf{1 1}^{2}$ and $\mathbf{1 5}^{2}$ in $99 \%$ yield with a ratio of $96: 4$ which was determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy. Treatment of erythro-11 (erythro: threo $=96: 4$ ) with benzyloxylamine hydrochloride and trimethylaluminium ${ }^{9}$ afforded the $O$-benzyl hydroxamate 12 (erythro:threo $=94: 6$ ) in $81 \%$ yield. $S$ Alkylation of the sulfide $\mathbf{1 2}$ (erythro: threo $=94: 6$ ) with methyl iodide in the presence of silver perchlorate ${ }^{4,10}$ followed by treatment of the resulting sulfonium salt with potassium carbonate caused it to undergo smooth lactamization by intra-
molecular substitution to give the trans-lactam 13 with high stereoselectivity (trans: cis $=90: 10$ ) in $60 \%$ yield. In a similar manner, the ( $Z$ )-ester $\mathbf{1 4}$ was transformed into the cis- $\beta$-lactam $\mathbf{1 7}$ via the threo-adduct $\mathbf{1 5}^{2}$ and the threo-hydroxamate $\mathbf{1 6}$ with high stereoselectivity (cis: trans $=92: 8$ ) (Scheme 3). The stereostructures of $\mathbf{1 3}$ and $\mathbf{1 7}$ were confirmed by their spectral data. The trans- $\beta$-lactam 13 showed a molecular ion peak at $\mathrm{m} / \mathrm{z} 205$, an IR absorption at $1752 \mathrm{~cm}^{-1}$ ( $\beta$-lactam) and ${ }^{1} \mathrm{H}$ NMR peaks at $\delta 2.46(1 \mathrm{H}, \mathrm{qd}, J 7.5,2,3-\mathrm{H})$ and $3.21(1 \mathrm{H}, \mathrm{qd}, J 6,2,4-\mathrm{H})$. Similarly, the cis- $\beta$-lactam 17 showed the following characteristic spectra $\left[\mathrm{m} / \mathrm{z} 205\left(\mathrm{M}^{+}\right), v_{\max } 1754 \mathrm{~cm}^{-1}\right.$ ( $\beta$-lactam), $\delta 2.92$ ( 1 H, qd, $J 7.5,6,3-\mathrm{H}$ ), $3.72(1 \mathrm{H}$, quint., $J 6,4-\mathrm{H})$ ]. It has been established ${ }^{11}$ that in the series of 3,4 -disubstituted azetidinones, the coupling constants ( $J$ values) between the 3 - and 4 hydrogens in the 3,4 -trans series ( $2.2-2.8 \mathrm{~Hz}$ ) are smaller than those in the cis series ( $5.0-5.9 \mathrm{~Hz}$ ). Their observation in the present cases showed a $J_{3,4}$ value of 2 Hz for 13 and a $J_{3,4}$ value of 6 Hz for $\mathbf{1 7}$. The result established that the $\beta$-lactam 13 is the trans-isomer and the $\beta$-lactam 17 is the cis-isomer. Thus, we have developed a new and simple synthetic method for making 3,4-disubstituted $\beta$-lactams.

## Synthesis of a key intermediate for carbapenem antibiotic (+)-PS-5

Carbapenem antibiotics PS-5, PS-6 and thienamycin comprise an interesting family of streptomycete metabolites characterized by the presence of the 7-oxo-1-azabicyclo[3.2.0]hept-2-ene2 -carboxylic acid system. ${ }^{8}$ The main strategies towards their

synthesis usually take place first with the preparation of an appropriately substituted monocyclic $\beta$-lactam 18 (a key intermediate) with the correct stereochemistry at the $\mathrm{C}_{3}$ - and $\mathrm{C}_{4}$-positions of the $\beta$-lactam ring, followed by chemical manipulations $\left(\mathbf{1 8} \longrightarrow \mathbf{1 9 )}\right.$ ) at the $\mathrm{N}_{1}$ - and $\mathrm{C}_{4}$-positions and subsequent ring closure $(\mathbf{1 9} \longrightarrow \mathbf{2 0})$ to form the bicyclic carbapenem system in the last steps of the synthesis (Scheme 4).


Since in the asymmetric synthesis of PS-5, the first step of stereoselective $\beta$-lactam formation is a crucial step, we focused our attention on exploring a synthetic route to PS-5 applying our newly found strategy established above.

Before approaching its asymmetric synthesis, we examined
the synthesis of the key racemic intermediate 9 for ( $\pm$ )-PS- 5 . The requisite $(E)$-olefin $\mathbf{2 4}$ was prepared by the conventional method involving the Wittig-Horner reaction and subsequent olefin isomerization (Scheme 5). The Swern oxidation of 3-


Scheme 5 Reagents: i, $(\mathrm{COCl})_{2}, \mathrm{DMSO} ; \mathrm{ii}, \mathrm{NaH}$; iii, $(\mathrm{PhS})_{2}$; iv, 0.1 equiv. $\mathrm{PhSLi}, 10$ equiv. PhSH ; v, $\mathrm{MeONH}_{2} \cdot \mathrm{HCl}, \mathrm{Me}_{3} \mathrm{Al}$
benzyloxypropanol 21 followed by the Wittig-Horner reaction of the resulting aldehyde $\mathbf{2 2}{ }^{12,13}$ with the phosphonate $\mathbf{2 3}^{14}$ gave a $1: 1$ mixture of $(E) \mathbf{- 2 4}$ and $(Z)-\mathbf{2 5}$ in $91 \%$ combined yield. The unstable $(Z)-\mathbf{2 5}$ was isomerized to the stable $(E)-\mathbf{2 4}$ upon heating at $100^{\circ} \mathrm{C}$ in the presence of diphenyl disulfide. ${ }^{15}(E)$ Ester 24 and the corresponding hydroxamate 27, which was readily prepared in $63 \%$ yield from the former by treatment with methoxylamine hydrochloride in the presence of trimethylaluminium, were subjected to the Michael-type addition of benzenethiol. ( $E$ )-24 was treated with benzenethiol in the presence of lithium benzenethiolate at room temperature to give a mixture of two adducts, ( $\pm$ )-erythro- 26 and its ( $\pm$ )-threo-isomer, as an inseparable mixture in $90 \%$ yield with a ratio of $85: 15$. The addition reaction of benzenethiol to the hydroxamate 27 proceeded smoothly at $60^{\circ} \mathrm{C}$ to give a mixture of two adducts, $( \pm)$-erythro- 28 and its ( $\pm$ )-threo-isomer, as an inseparable mixture in $83 \%$ yield with high stereoselectivity (erythro: threo $=91: 9) \quad($ Scheme 5). Unfortunately, treatment of the $( \pm$ )-erythro-ester 26 (erythro:threo $=85: 15$ ) with methoxylamine hydrochloride in the presence of trimethylaluminium gave the corresponding $O$-methyl ( $\pm$ )-hydroxamate 28 (erythro: threo $=86: 14)$ in only $10 \%$ yield as an isolated product in addition to a complex mixture. The structure of the ( $\pm$ )-erythroadduct $\mathbf{2 8}$ was determined by the facts that it showed a molecular ion peak at $m / z 373$ and ${ }^{1} \mathrm{H}$ NMR peaks at $\delta 2.18(1 \mathrm{H}, \mathrm{m}$, $2-\mathrm{H})$ and $3.54(1 \mathrm{H}, \mathrm{td}, J 7.5,4,3-\mathrm{H})$. The relative configuration of the ( $\pm$ )-erythro- $\mathbf{2 8}$ was established by its chemical conversion into the $( \pm$-trans- $\beta$-lactam 29 (trans:cis $=95: 5$ ) which was accomplished in $83 \%$ yield through $S$-alkylation of ( $\pm$ )-28 (erythro: threo $=91: 9$ ) and subsequent treatment of the resulting sulfonium salt with potassium carbonate (Scheme 6). The ( $\pm$ )-trans-azetidinone 29 thus prepared showed a molecular ion


Scheme 6 Reagents: i, 1) MeI, $\left.\mathrm{AgClO}_{4}, 2\right) \mathrm{K}_{2} \mathrm{CO}_{3}$; ii, Ca , liq. $\mathrm{NH}_{3}$; iii, TBDMSOTf, 2,6-lutidine
peak at $m / z 263$, an IR absorption at $1760 \mathrm{~cm}^{-1}$ ( $\beta$-lactam) and ${ }^{1} \mathrm{H}$ NMR peaks at $\delta 2.57(1 \mathrm{H}$, ddd, $J 8,6,2,3-\mathrm{H})$ and $3.71(1 \mathrm{H}$, ddd, J 7.5, 6, 2, 4-H).

Next, $( \pm)$-29 was converted into ( $\pm$ )-9, a known ${ }^{13}$ key intermediate for the synthesis of ( $\pm$ )-PS-5. Reductive cleavage of both the $\mathrm{N}-\mathrm{O}$ and $\mathrm{O}-\mathrm{CH}_{2} \mathrm{Ph}$ bonds of $( \pm)-29$ by use of calcium ${ }^{17,18}$ in liquid ammonia proceeded smoothly to give the desired product ( $\pm$ )-30 in $96 \%$ yield whilst the conventional method, ${ }^{13,16}$ using sodium metal ( $\mathrm{Na} /$ liq. $\mathrm{NH}_{3},-78^{\circ} \mathrm{C}$ ), was unsuccessful. Finally, treatment of ( $\pm$ )-30 with TBDMSOTf in the presence of 2,6 -lutidine afforded the disilylated $( \pm)-\beta$ lactam 9, which was identical with an authentic specimen on comparison of their spectral data (Scheme 6). ${ }^{13,19}$
The above result established that a new synthetic route to PS5 based on the racemic compound ( $\pm$ ) -9 was successful and was then applied to the asymmetric synthesis of $(-)-\beta$-lactam 9 . We first examined the addition of benzenethiol to the olefins having oxazolidinone ${ }^{20}$ as a chiral auxiliary (Scheme 7, Table 1).


Scheme 7
Unsaturated chiral imide 33 was readily prepared by condensation of the acid chloride 31b with the lithiated oxazolidinone. The imide $(E)-33$ was treated with 10 equiv. of benzenethiol in the presence of 1 equiv. of lithium benzenethiolate in THF at $-78^{\circ} \mathrm{C}$ to give a mixture of two products $\left(2^{\prime} S, 3^{\prime} S\right)-34$ and

Table 1 Addition reaction of benzenethiol to the chiral imide 33

|  | PhSLi/PhSH <br> (equiv.) | Time (h) | Solv. | Yield (\%) | Ratio <br> $\mathbf{3 4 : 3 5}$ |
| :--- | :--- | :--- | :--- | :---: | :---: |
| 1 | $1 / 10$ | 5 | THF | 33 | $74: 26$ |
| 2 | $1 / 10$ | 5 | THF | 74 | $75: 25$ |
|  | $[+$ LiI (1.3 equiv.)] |  |  |  |  |
| 3 | $5 / 10$ | 5 | THF | 97 | $80: 20$ |
| 4 | $1 / 10$ | 0.6 | $\mathrm{Et}_{2} \mathrm{O}$ | 100 | $50: 50$ |

$\left(2^{\prime} R, 3^{\prime} R\right)$ - $\mathbf{3 5}$ in $33 \%$ yield with a ratio of 74:26 (entry 1 ). These diastereomers were separated by medium pressure column chromatography. In the presence of lithium iodide as an additive, a mixture of two adducts $\mathbf{3 4}$ and $\mathbf{3 5}$ was obtained in $74 \%$ yield with almost the same diastereoselectivity (entry 2) Employment of 5 equiv. of lithium benzenethiolate improved the yield and the ratio of a mixture of $\mathbf{3 4}$ and $\mathbf{3 5}$ to $97 \%$ yield and $80: 20$, respectively (entry 3 ). The addition reaction in diethyl ether proceeded very smoothly but the stereoselectivity was markedly diminished giving a $1: 1$ mixture of $\mathbf{3 4}$ and 35 (entry 4). The structures of the adducts 34 and 35 were determined based on the facts that these adducts showed a molecular ion peak at $m / z 455$ and ${ }^{1} \mathrm{H}$ NMR peaks due to the 2- and 3hydrogens as shown in the Experimental section. The absolute configuration of ( $2^{\prime} S, 3^{\prime} S$ )-34 was established by its chemical conversion into the known ${ }^{13,19}$ intermediate ( - )-9 for the synthesis of (+)-PS-5. The absolute configuration of the minor adduct ( $2^{\prime} R, 3^{\prime} R$ )-35 was determined as follows (Scheme 8 ). As


Scheme 8
described later, $\left(2^{\prime} S, 3^{\prime} S\right)$ - $\mathbf{3 4}$ and $\left(2^{\prime} R, 3^{\prime} R\right)$ - $\mathbf{3 5}$ were converted into the enantiomeric thioesters $(2 R, 3 S)-36$ and $(2 S, 3 R)-37$, respectively which showed identical IR and ${ }^{1} \mathrm{H}$ NMR spectra. Comparison of the optical rotation of $(2 S, 3 R)$ - 37 with that of $(2 R, 3 S)$ - $\mathbf{3 6}$ established unambiguously their absolute configurations.

We propose the possible reaction pathway for the addition of benzenethiol to 33 in THF as shown in Scheme $9 .{ }^{2 b, 3 b, 21}$ It is presumed that both carbonyl groups of the $\alpha, \beta$-unsaturated carboxylic acid moiety and the chiral auxiliary would be fixed by chelation with lithium benzenethiolate. The conformation of the chiral imide with respect to the rotamers arising from rotation about the $\mathrm{C}-1^{\prime}-\mathrm{C}-2^{\prime}$ axis would be in the s -trans form C due to steric hindrance between the ethyl and the isopropyl groups on the chiral auxiliary found in the corresponding s -cis conformation D. Addition of lithium benzenethiolate to the metal coordinated imide $\mathbf{C}$ would occur from the $\beta$-face, according to the 1,5 -asymmetric induction by the isopropyl group on the oxazolidinone ring, to form the enolate $\mathbf{E}$. The following protonation of $\mathbf{E}$ would occur from the $\alpha$-face, due to the stereoelectronic effect ${ }^{2 b}$ of the newly introduced sulfur group in overcoming the steric hindrance of the isopropyl group, to give ( $2^{\prime} S, 3^{\prime} S$ )-34.

We then investigated conversion of $\left(2^{\prime} S, 3^{\prime} S\right)$ - $\mathbf{3 4}$ into $(2 S, 3 S)$ 28 via cleavage of the chiral auxiliary (Scheme 10). We first attempted transamination of $\mathbf{3 4}$ into hydroxamate ( $2 S, 3 S$ )-28 in one step by treatment with methoxylamine hydrochloride


Scheme 9

$(2 R, 3 S)-\mathbf{3 6}$
$(2 S, 3 S)-\mathbf{3 9}$
Scheme 10 Reagents: i, $\mathrm{MeONH}_{2} \cdot \mathrm{HCl}, \mathrm{Me}_{3} \mathrm{Al}$; ii, $\mathrm{Bu}^{n} \mathrm{Li}, \mathrm{BnSH}$, $\mathrm{Me}_{3} \mathrm{Al}$; iii, $\mathrm{CF}_{3} \mathrm{CO}_{2} \mathrm{Ag}, \mathrm{H}_{2} \mathrm{O}$; iv, $\mathrm{MeONH}_{2} \cdot \mathrm{HCl}$, WSC
and trimethylaluminium. ${ }^{9,22}$ However, the desired hydroxamate ( $2 S, 3 S$ )-28 was obtained in only $5 \%$ yield, although the undesired amide 38 was obtained in $84 \%$ yield as a result of the attack of methoxylamine onto the oxazolidinone carbonyl group. Recently, we have found ${ }^{4}$ that the aluminium thiobenzyloxy 'ate' complex is an excellent reagent for the cleavage of N -acyloxazolidinones and camphor sultams. Therefore, this method was applied to $\left(2^{\prime} S, 3^{\prime} S\right)-34$. As expected, removal of the chiral auxiliary by treatment with the aluminium thiobenzyloxy 'ate' complex was smoothly achieved to give the thioester $(2 R, 3 S)-\mathbf{3 6}$ in $97 \%$ yield, which was also obtained in $63 \%$ yield by a known ${ }^{23}$ approach employing lithium phenylmethanethiolate as a nucleophile. Hydrolysis of the thioester $(2 R, 3 S)-\mathbf{3 6}$ in the presence of silver trifluoroacetate proceeded smoothly to give the corresponding acid $(2 S, 3 S)-39$ in $96 \%$ yield which was also obtained, but in only $45 \%$ yield, in the presence of silver perchlorate. ${ }^{4}(2 S, 3 S)$ - $\mathbf{3 9}$ was then treated with methoxylamine hydrochloride in the presence of 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (WSC) to give the desired hydroxamate $(2 S, 3 S)-\mathbf{2 8}$ in $86 \%$ yield. According to the procedure established with the racemic $( \pm)-9$, formation of the azetidinone ( $3 R, 4 R$ )-29, cleavage of both the $\mathrm{N}-\mathrm{O}$ and $\mathrm{O}-\mathrm{CH}_{2} \mathrm{Ph}$ bonds of $(3 R, 4 R)-29$ and subsequent disilylation afforded the disilylated $\beta$-lactam $(3 R, 4 R)-9,[a]_{\mathrm{D}}^{26}-34.2\left(c 0.79, \mathrm{CHCl}_{3}\right)\left\{\right.$ lit., ${ }^{19}$ $[a]_{\mathrm{D}}-37.73\left(c 2.25, \mathrm{CHCl}_{3}\right) ;$ lit., $\left.{ }^{13}[a]_{\mathrm{D}}-30.1\left(c 2.9, \mathrm{CHCl}_{3}\right)\right\}$ in
$42 \%$ overall yield from the carboxylic acid 31a in a nine-step sequence (Scheme 6). Since $(3 R, 4 R)-9$ had previously been transformed into (+)-PS-5, ${ }^{24-27}$ the present method provides a new highly efficient asymmetric synthesis of $(+)$-PS- 5 .

In conclusion, we have now developed a new strategy for the stereoselective construction of the $\beta$-lactam ring via stereospecific nucleophilic addition of thiols and the stereoselective displacement of the corresponding sulfonium group with $O$ alkyl hydroxamate anions. This method has been successfully applied to the asymmetric synthesis of a key intermediate for the synthesis of carbapenem antibiotic (+)-PS-5.

## Experimental

${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were measured using Varian Gemini$200\left({ }^{1} \mathrm{H}, 200 \mathrm{MHz}\right)$, Varian Gemini-300 $\left({ }^{1} \mathrm{H}, 300 \mathrm{MHz}\right)$ and VXR-500 $\left({ }^{1} \mathrm{H}, 500 \mathrm{MHz} ;{ }^{13} \mathrm{C}, 125 \mathrm{MHz}\right)$ instruments for solutions in deuteriochloroform, unless otherwise stated (tetramethylsilane was used as the internal reference); $J$ values are given in Hz. IR spectra were measured with a Perkin-Elmer 1600 FTIR machine for solutions in chloroform, unless otherwise stated and mass spectra were taken with an Hitachi M-4100 spectrometer. Mps were determined with a Kofler-type hot-stage apparatus and are uncorrected. All reactions were performed under nitrogen and extracts from the reaction mixtures were washed with water, dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated under reduced pressure. TLC was performed on precoated silica gel $60 \mathrm{~F}_{254}(0.25 \mathrm{~mm}$ thick, Merck) with UV detection at 254 and 300 nm . Medium-pressure column chromatography (MPCC) was undertaken on a $530-4-10 \mathrm{~V}$ apparatus (Yamazen) with Lobar größe B (310-25, Lichroprep Si60, Merck) as column absorbent. For flash column chromatography (FCC), Merck Kieselgel 60 (230-400 mesh) was used. Short column chromatography (SCC) was undertaken on a short glass filter using Merck Kieselgel 60 (230-400 mesh) under reduced pressure. For inseparable mixtures, the ${ }^{1} \mathrm{H}$ NMR spectra of the major isomer is assigned.
erythro-2-Methyl-N-benzyloxy-3-(phenylsulfanyl)butanamide 12 According to the method previously described, ${ }^{2}$ the erythroester $\mathbf{1 1}^{\mathbf{2}}$ (erythro: threo $=96: 4$ ) was prepared by the addition of benzenethiol to $(E)$-methyl 2-methylbut-2-enoate. Trimethylaluminium ( 2 m solution in hexane; $1.2 \mathrm{~cm}^{3}, 2.4 \mathrm{mmol}$ ) was added with stirring at $0^{\circ} \mathrm{C}$ to a suspension of $O$-benzylhydroxylamine hydrochloride ( $382 \mathrm{mg}, 2.4 \mathrm{mmol}$ ) in toluene $\left(2.4 \mathrm{~cm}^{3}\right)$. This resulting solution was added with stirring at room temperature to a solution of the erythro-ester $11(269 \mathrm{mg}$, $1.2 \mathrm{mmol})$ in toluene $\left(4 \mathrm{~cm}^{3}\right)$. After being stirred at $60{ }^{\circ} \mathrm{C}$ for 1 h , the mixture was acidified with $10 \%$ hydrochloric acid and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was washed, dried and concentrated to give a residue which was purified by MPCC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to give a mixture of the erythro-hydroxamate 12 and its threo-isomer (erythro: threo $=94: 6)(306 \mathrm{mg}, 81 \%)$ as a colorless oil. The ratio of erythro- to threo-adducts was determined by $200 \mathrm{MHz}^{1} \mathrm{H}$ NMR spectroscopy. $v_{\text {max }} / \mathrm{cm}^{-1} 3398$ (NH) and 1690 (NCO) (Found: $\mathrm{M}^{+}$, 315.1290. $\mathrm{C}_{18} \mathrm{H}_{21} \mathrm{NO}_{2} \mathrm{~S}$ requires $M, 315.1291)$. erythro-12; $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 8.44(1 \mathrm{H}, \mathrm{s}$, $\mathrm{NH}), 7.50-7.22(10 \mathrm{H}, \mathrm{m}, \mathrm{Ph} \times 2), 4.95\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{Ph}\right), 3.38$ $(1 \mathrm{H}$, quint., $J 7,3-\mathrm{H}), 2.19(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H})$ and $1.28(6 \mathrm{H}, \mathrm{d}, J 7$, $\mathrm{Me} \times 2$ ) .

## trans-3,4-Dimethyl-1-(benzyloxy)azetidin-2-one 13

Methyl iodide ( $2 \mathrm{~cm}^{3}, 31 \mathrm{mmol}$ ) was added with stirring at room temperature to a solution of the erythro- $\mathbf{1 2}(95 \mathrm{mg}, 0.3 \mathrm{mmol})$ (erythro: threo $=94: 6)$ and silver perchlorate $(280 \mathrm{mg}, 1.4$ $\mathrm{mmol})$ in $\mathrm{MeCN}\left(4 \mathrm{~cm}^{3}\right)$. After being stirred at room temperature for 15 h , the mixture was filtered to remove the resulting silver iodide. The filtrate was concentrated to give the sulfonium salt. To a stirred suspension of $\mathrm{K}_{2} \mathrm{CO}_{3}(360 \mathrm{mg}, 2.6 \mathrm{mmol})$ in refluxing acetone $\left(12 \mathrm{~cm}^{3}\right)$ was added a solution of the sulf-
onium salt in acetone $\left(7 \mathrm{~cm}^{3}\right)$ and the mixture was refluxed for 2 h . After addition of water, the mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The extract was dried and concentrated to give the residue which was purified by $\operatorname{MPCC}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{AcOEt} 10: 1\right)$ to give the trans-azetidinone 13 (trans: cis $=90: 10$ ) as a colorless oil ( $37 \mathrm{mg}, 60 \%$ ). The ratio of trans- to cis-products was determined by $200 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectroscopy. $v_{\max } / \mathrm{cm}^{-1} 1752$ (NCO) (Found: $\mathrm{M}^{+}$, 205.1105. $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{NO}_{2}$ requires $M$, 205.1102). trans-13; $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.46(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 5.02$ and $4.98\left(2 \mathrm{H}, \mathrm{ABq}, J 12, \mathrm{OCH}_{2} \mathrm{Ph}\right), 3.21(1 \mathrm{H}, \mathrm{qd}, J 6,2,4-\mathrm{H}), 2.46$ $(1 \mathrm{H}, \mathrm{qd}, J 7.5,2,3-\mathrm{H}), 1.24(3 \mathrm{H}, \mathrm{d}, J 7.5,3-\mathrm{Me})$ and $1.19(3 \mathrm{H}$, d, $J 6,4-\mathrm{Me}$ ).

## threo-2-Methyl- $N$-benzyloxy-3-(phenylsulfanyl)butanamide 16

According to the method previously described, ${ }^{2}$ the threo-ester $\mathbf{1 5}^{\mathbf{2}}$ (threo: erythro $=86: 14$ ) was prepared by the addition of benzenethiol to $(Z)$-methyl 2-methylbut-2-enoate. According to the procedure described for the preparation of erythrohydroxamate 12, a solution of threo- $\mathbf{1 5}$ (threo : erythro $=86: 14$ ) $(134 \mathrm{mg}, 0.6 \mathrm{mmol}), O$-benzylhydroxylamine hydrochloride and trimethylaluminium in toluene was stirred at $60^{\circ} \mathrm{C}$ for 3 h . The crude product was purified by MPCC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to give a mixture of the threo-hydroxamate $\mathbf{1 6}$ and the erythro-isomer (threo: erythro $=85: 15)(155 \mathrm{mg}, 82 \%)$ as a colorless oil. The ratio of erythro- to threo-adducts was determined by 200 MHz ${ }^{1} \mathrm{H}$ NMR spectroscopy. $v_{\max } / \mathrm{cm}^{-1} 3398$ (NH) and 1692 (NCO) (Found: $\mathrm{M}^{+}$, 315.1277. $\mathrm{C}_{18} \mathrm{H}_{21} \mathrm{NO}_{2} \mathrm{~S}$ requires $M, 315.1291$ ). threo-16; $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 8.84(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 7.50-7.20(10 \mathrm{H}, \mathrm{m}$, $\mathrm{Ph} \times 2), 4.91\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{Ph}\right), 3.46(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 2.16(1 \mathrm{H}, \mathrm{m}$, $2-\mathrm{H}), 1.24$ and 1.12 (each $3 \mathrm{H}, \mathrm{d}, J 7, \mathrm{Me} \times 2$ ).

## cis-3,4-Dimethyl-1-(benzyloxy)azetidin-2-one 17

According to the procedure described for the preparation of trans-13, the threo-16 (threo: erythro $=85: 15)(190 \mathrm{mg}, 0.6$ mmol ) was alkylated with methyl iodide in the presence of silver perchlorate and the resulting sulfonium salt was treated with $\mathrm{K}_{2} \mathrm{CO}_{3}$. The crude product was purified by MPCC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to give the cis-azetidinone $\mathbf{1 7}$ (cis:trans $=92: 8$ ) as a colorless oil ( $71 \mathrm{mg}, 58 \%$ ). The ratio of trans- to cis-products was determined by $200 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectroscopy. $v_{\text {max }} / \mathrm{cm}^{-1}$ 1754 (NCO) (Found: $\mathrm{M}^{+}, 205.1092 . \mathrm{C}_{12} \mathrm{H}_{15} \mathrm{NO}_{2}$ requires $M$, 205.1102). cis-17; $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.44(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 4.98(2 \mathrm{H}$, $\left.\mathrm{s}, \mathrm{OCH}_{2} \mathrm{Ph}\right), 3.72(1 \mathrm{H}$, quint., $J 6,4-\mathrm{H}), 2.92(1 \mathrm{H}, \mathrm{qd}, J 7.5,6$, $3-\mathrm{H}), 1.10(3 \mathrm{H}, \mathrm{d}, J 7.5,3-\mathrm{Me})$ and $1.03(3 \mathrm{H}, \mathrm{d}, J 6,4-\mathrm{Me})$.

## ( $E$ )-Ethyl 2-ethyl-5-(benzyloxy)pent-2-enoate 24

According to the literature procedure, ${ }^{12,13}$ aldehyde $\mathbf{2 2}^{12,13}$ was prepared from alcohol 21. A solution of triethyl $\alpha$ phosphonopropionate $23{ }^{14}$ in THF $\left(50 \mathrm{~cm}^{3}\right)$ was added at $0{ }^{\circ} \mathrm{C}$ to a stirred suspension of $\mathrm{NaH}(60 \%$ dispersion in mineral oil; $0.8 \mathrm{~g}, 20 \mathrm{mmol})$ in THF $\left(10 \mathrm{~cm}^{3}\right)$. After being stirred at room temperature for 0.5 h , a solution of the aldehyde $22(3.28 \mathrm{~g}, 20$ mmol ) in THF ( $50 \mathrm{~cm}^{3}$ ) was added dropwise at $0{ }^{\circ} \mathrm{C}$ and the whole was stirred at the same temperature for 1 h . The resulting solution was poured into ice-cooled water $\left(20 \mathrm{~cm}^{3}\right)$ and then extracted with $\mathrm{Et}_{2} \mathrm{O}$. The extract was washed with saturated aqueous NaCl , dried and concentrated to give a mixture of $(E)$ ester 24 and $(Z)$-ester $\mathbf{2 5}(\mathbf{2 4 : 2 5}=1: 1)$ as a colorless oil ( 4.8 g , $91 \%$ ). The ratio of $(E)$ - to $(Z)$-esters was determined by 200 $\mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectroscopy. $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.40(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph})$, $6.79[0.5 \mathrm{H}, \mathrm{t}, J 7,(E)-3-\mathrm{H}], 6.00[0.5 \mathrm{H}, \mathrm{t}, J 7,(Z)-3-\mathrm{H}], 4.59$ and 4.58 (each $\left.1 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{Ph}\right), 4.24\left(2 \mathrm{H}, \mathrm{m}, \mathrm{OCH}_{2} \mathrm{Me}\right), 3.60[1 \mathrm{H}, \mathrm{t}$, $\left.J 7,(E)-5-\mathrm{H}_{2}\right], 3.58\left[1 \mathrm{H}, \mathrm{t}, J 7,(Z)-5-\mathrm{H}_{2}\right], 2.80(1 \mathrm{H}), 2.56(1 \mathrm{H})$ and $2.36(2 \mathrm{H})$ (each q, $J 7,4-\mathrm{H}_{2}$ and $\left.2-\mathrm{CH}_{2}\right)$ and $1.40-0.96(6 \mathrm{H}$, $\mathrm{m}, M e \times 2)$. A solution of a mixture of $(E)$ - and $(Z)$-esters 24 and $25(3.6 \mathrm{~g}, 14 \mathrm{mmol})$ and diphenyl disulfide ${ }^{15}(914 \mathrm{mg}, 4.2$ $\mathrm{mmol})$ in THF ( $170 \mathrm{~cm}^{3}$ ) was refluxed for 3 h . The mixture was then cooled and the solvent was removed under reduced pressure. The residue was purified by MPCC ( $n$-hexane-AcOEt $15: 1)$ to give the $(E)$-ester 24 as a colorless oil ( $3.5 \mathrm{~g}, 95 \%$ ).
$v_{\max } / \mathrm{cm}^{-1} 1702(\mathrm{COOEt}) ; m / z(\mathrm{CI}) 263\left(\mathrm{MH}^{+}\right) ; \delta_{\mathrm{H}}(200 \mathrm{MHz})$ $7.40(5 \mathrm{H}, \mathrm{s}, \mathrm{Ph}), 6.79(1 \mathrm{H}, \mathrm{t}, J 7,3-\mathrm{H}), 4.58\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{Ph}\right)$, $4.24\left(2 \mathrm{H}, \mathrm{q}, J 7, \mathrm{OCH}_{2} \mathrm{Me}\right), 3.60\left(2 \mathrm{H}, \mathrm{t}, J 7,5-\mathrm{H}_{2}\right), 2.56$ and 2.36 (each $2 \mathrm{H}, \mathrm{q}, J 7,4-\mathrm{H}_{2}$ and $\left.2-\mathrm{CH}_{2}\right), 1.32(3 \mathrm{H}, \mathrm{t}, J 7$, $\left.\mathrm{OCH}_{2} \mathrm{Me}\right)$ and $1.02\left(3 \mathrm{H}, \mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{Me}\right)$.

## ( $\pm$ )-Ethyl erythro-2-ethyl-5-benzyloxy-3-(phenylsulfanyl)pentanoate 26

Benzenethiol ( $0.38 \mathrm{~cm}^{3}, 3.7 \mathrm{mmol}$ ) was added with stirring at $0^{\circ} \mathrm{C}$ to a solution of $n$-butyllithium ( 1.63 m solution in hexane; $\left.0.02 \mathrm{~cm}^{3}, 0.037 \mathrm{mmol}\right)$ in THF $\left(2.5 \mathrm{~cm}^{3}\right)$ to give a solution of a 10:0.1 mixture of benzenethiol and lithium benzenethiolate. To the resulting solution was added a solution of $\mathbf{2 4}(97 \mathrm{mg}, 0.37$ mmol ) in THF ( $2.5 \mathrm{~cm}^{3}$ ) at room temperature. After being stirred at room temperature for 2 h , the mixture was made alkaline by adding $5 \%$ aqueous NaOH and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The extract was washed, dried and concentrated to give the residue which was purified by MPCC ( $n$-hexane$\left.\mathrm{CH}_{2} \mathrm{Cl}_{2} 1: 1\right)$ to give a mixture of ( $\pm$ )-erythro- 26 and the threoisomer (erythro: threo $=85: 15)(124 \mathrm{mg}, 90 \%)$ as a colorless oil. The ratio of erythro- to threo-adducts was determined by $200 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectroscopy. $v_{\text {max }} / \mathrm{cm}^{-1} 1724\left(\mathrm{CO}_{2} \mathrm{Et}\right)$ (Found: $\mathrm{M}^{+}$, 372.1758. $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{O}_{3} \mathrm{~S}$ requires $M, 372.1758$ ) erythro-26; $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.50-7.20(10 \mathrm{H}, \mathrm{m}, \mathrm{Ph} \times 2), 4.48(2 \mathrm{H}$, $\left.\mathrm{s}, \mathrm{OCH}_{2} \mathrm{Ph}\right), 4.10\left(2 \mathrm{H}, \mathrm{m}, \mathrm{OCH}_{2} \mathrm{Me}\right), 3.82-3.60\left(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}_{2}\right)$, $3.51(1 \mathrm{H}$, ddd, $J 11.5,8,4.5,3-\mathrm{H}), 2.50(1 \mathrm{H}, \mathrm{ddd}, J 10,8,4.5$, $2-\mathrm{H}), 2.16-1.66\left(4 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}\right.$ and $\left.4-\mathrm{H}_{2}\right), 1.22(3 \mathrm{H}, \mathrm{t}, J 7$, $\left.\mathrm{OCH}_{2} \mathrm{Me}\right)$ and $0.88\left(3 \mathrm{H}, \mathrm{t}, \mathrm{J}, \mathrm{CH}_{2} \mathrm{Me}\right)$.

## ( $\pm$ )-erythro-2-Ethyl- N -methoxy-5-benzyloxy-3-(phenylsulfanyl)pentanamide 28

According to the procedure described for the preparation of the erythro-hydroxamate 12, a solution of ( $\pm$ )-erythro- $\mathbf{2 6}$ (erythro: threo $=85: 15)(74 \mathrm{mg}, 0.2 \mathrm{mmol})$, methoxylamine hydrochloride and trimethylaluminium in toluene $\left(1.5 \mathrm{~cm}^{3}\right)$ was stirred under reflux for 3 h . The crude product was purified by MPCC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}-n\right.$-hexane 1:1) to give a mixture of the ( $\pm$ )-erythro-hydroxamate $\mathbf{2 8}$ and the threo-isomer (erythro:threo $=$ $86: 14)(7 \mathrm{mg}, 10 \%)$ as a colorless oil. The ratio of erythro- to threo-adducts was determined by $200 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectroscopy. $v_{\text {max }} / \mathrm{cm}^{-1} 3400(\mathrm{NH}), 1694$ (NCO) (Found: $\mathrm{M}^{+}$, 373.1703. $\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{NO}_{3} \mathrm{~S}$ requires $\left.M, 373.1701\right)$. 28; $\delta_{\mathrm{H}}(200 \mathrm{MHz})$ $8.70(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{NH}), 7.44-7.30(10 \mathrm{H}, \mathrm{m}, \mathrm{Ph} \times 2), 4.52$ and 4.46 $\left(2 \mathrm{H}, \mathrm{ABq}, J 11, \mathrm{OCH}_{2} \mathrm{Ph}\right), 3.90(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 3.66(1 \mathrm{H}, \mathrm{m}$, $5-\mathrm{H}), 3.64(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.54(1 \mathrm{H}, \mathrm{td}, J 7.5,4,3-\mathrm{H}), 2.18(1 \mathrm{H}$, $\mathrm{m}, 2-\mathrm{H}), 1.99(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 1.91-1.76\left(3 \mathrm{H}, \mathrm{m}, 2-\mathrm{CH}_{2}\right.$ and $\left.4-\mathrm{H}\right)$ and $0.86\left(3 \mathrm{H}, \mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{Me}\right)$.

## ( $E$ )-2-Ethyl- $N$-methoxy-5-(benzyloxy)pent-2-enamide 27

According to the procedure described for the preparation of the hydroxamate 12, a solution of the ester 24 ( $286 \mathrm{mg}, 1.1 \mathrm{mmol}$ ), methoxylamine hydrochloride and trimethylaluminium in toluene was stirred at $60^{\circ} \mathrm{C}$ for 5 h . The crude product was purified by MPCC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{AcOEt} 5: 1\right)$ to give the $(E)$ hydroxamate $27(182 \mathrm{mg}, 63 \%)$ as a colorless oil; $v_{\text {max }} / \mathrm{cm}^{-1}$ 3404 (NH), 1680 (NCO) (Found: $\mathrm{M}^{+}$, 263.1507. $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{NO}_{3}$ requires $M, 263.2519)$; $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 8.59(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{NH}), 7.44$ $7.30(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 6.12(1 \mathrm{H}, \mathrm{t}, J 7,3-\mathrm{H}), 4.54\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH} \mathrm{H}_{2} \mathrm{Ph}\right)$, $3.79(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.56\left(2 \mathrm{H}, \mathrm{t}, J 7,5-\mathrm{H}_{2}\right), 2.46$ and 2.32 (each $2 \mathrm{H}, \mathrm{q}, J 7,4-\mathrm{H}_{2}$ and $\left.2-\mathrm{CH}_{2}\right)$ and $1.00\left(3 \mathrm{H}, \mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{Me}\right)$.

## Addition of benzenethiol to hydroxamate 27

According to the procedure described for the preparation of $( \pm)$-erythro-26, a solution of $27(145 \mathrm{mg}, 0.55 \mathrm{mmol}), \mathrm{PhSH}$ ( 5.5 mmol ) and $\operatorname{PhSLi}(0.055 \mathrm{mmol})$ in THF ( $3 \mathrm{~cm}^{3}$ ) was stirred under reflux for 2 h . The crude product was purified by MPCC ( $n$-hexane- $\mathrm{CH}_{2} \mathrm{Cl}_{2} 1: 1$ ) to give a mixture of ( $\pm$ )-erythro- 28 and its threo-isomer (erythro: threo $=91: 9)(170 \mathrm{mg}, 82 \%)$ as a colorless oil. The ratio of erythro- to threo-adducts was determined by $200 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectroscopy. The ( $\pm$ )-erythro- 28 was
identical with the authentic sample, prepared from ( $\pm$ )-erythro26 based on a comparison of their spectral data.

## ( $\pm$ )-trans-3-Ethyl-1-methoxy-4-[2-(benzyloxy)ethyl]azetidin-2one 29

According to the procedure described for the preparation of the trans-13, ( $\pm$ )-erythro-28 (erythro:threo $=91: 9)(116 \mathrm{mg}, 0.3$ mmol ) was alkylated with methyl iodide in the presence of silver perchlorate and then the resulting sulfonium salt was treated with $\mathrm{K}_{2} \mathrm{CO}_{3}$. The crude product was purified by MPCC (AcOEt- $\mathrm{CH}_{2} \mathrm{Cl}_{2} \quad 1: 10$ ) to give the ( $\pm$ )-trans-azetidinone 29 (trans:cis $95: 5$ ) as a colorless oil ( $66 \mathrm{mg}, 83 \%$ ). The ratio of trans- to cis-products was determined by $200 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectroscopy. $v_{\text {max }} / \mathrm{cm}^{-1} 1760$ (NCO) (Found: $\mathrm{M}^{+}, 263.1549$. $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{NO}_{3}$ requires $M, 263.1521$ ) (Found: C, 66.8; H, 7.9; N, 5.1. $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{NO}_{3} \cdot \frac{1}{3} \mathrm{H}_{2} \mathrm{O}$ requires $\left.\mathrm{C}, 66.9 ; \mathrm{H}, 8.1 ; \mathrm{N}, 5.2 \%\right)$. ( $\pm$ )-trans-29; $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.40-7.26(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 4.50(2 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{OCH}_{2} \mathrm{Ph}\right), 3.75(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.71(1 \mathrm{H}, \mathrm{ddd}, J 7.5,6,2,4-\mathrm{H})$, $3.60\left(2 \mathrm{H}, \mathrm{m}, \mathrm{OCH}_{2}\right), 2.57(1 \mathrm{H}$, ddd, $J 8,6,2,3-\mathrm{H}), 2.14(1 \mathrm{H}$, dq, $J 14,6,4-\mathrm{CH}), 1.91(1 \mathrm{H}, \mathrm{dtd}, J 14,7.5,6,4-\mathrm{CH}), 1.80-1.50$ $\left(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{CH}_{2}\right)$ and $1.00(3 \mathrm{H}, \mathrm{t}, J 7.5, \mathrm{Me}) ; \delta_{\mathrm{C}} 166.2(\mathrm{~s}), 138.0$ (s), 128.4 (d), 127.8 (d), 127.7 (d), 73.3 (t), 66.9 (t), 63.6 (q), 60.7 (d), 52.6 (d), 32.9 (t), 21.3 (t), 11.4 (q).

## ( $\pm$ )-trans-3-Ethyl-1-[(1,1-dimethylethyl)dimethylsilyl]-4-\{2-

 [(1,1-dimethylethyl)dimethylsilyloxy]ethyl\}azetidin-2-one 9Metallic calcium ( $58 \mathrm{mg}, 1.45 \mathrm{mmol}$ ) was added to liquid ammmonia $\left(10 \mathrm{~cm}^{3}\right)$ at $-78^{\circ} \mathrm{C}$ with stirring. A solution of $( \pm)$ 29 (trans: cis $=95: 5)(50 \mathrm{mg}, 0.19 \mathrm{mmol})$ in THF $\left(0.3 \mathrm{~cm}^{3}\right)$ was added to the above solution and the resulting blue-colored solution was stirred for 2 h . Ammonium chloride was added until the blue color disappeared, then the liquid ammonia was evaporated to give a residue, to which water was added. The resulting mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The extract was washed, dried and concentrated to give the ( $\pm$ )-alcohol 30. $v_{\text {max }} /$ $\mathrm{cm}^{-1} 3500-3400(\mathrm{NH}, \mathrm{OH})$ and $1760(\mathrm{NCO}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz})$ 3.90-3.64 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{OCH}_{2}$ ), 3.48 ( 1 H , ddd, J 8, 5.5, 2, 4-H), 2.78 $(1 \mathrm{H}$, br t, $J 8,3-\mathrm{H}), 2.00-1.50\left(4 \mathrm{H}, \mathrm{m}, 4-\mathrm{CH}_{2}\right.$ and $\left.3-\mathrm{CH}_{2}\right)$ and $1.02(3 \mathrm{H}, \mathrm{t}, J 8, \mathrm{Me})$.
According to the literature, ${ }^{13,19}$ a solution of the $( \pm)$-alcohol $30(15 \mathrm{mg}, 0.1 \mathrm{mmol})$, , 2, 6 -lutidine ( $0.12 \mathrm{~cm}^{3}, 1.1 \mathrm{mmol}$ ) and TBDMSOTf ( $0.09 \mathrm{~cm}^{3}, 0.4 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(5 \mathrm{~cm}^{3}\right)$ was stirred at $0^{\circ} \mathrm{C}$ for 1 h . After addition of $\mathrm{MeOH}\left(0.25 \mathrm{~cm}^{3}\right)$, the mixture was concentrated to give the residue which was purified by FCC ( $n$-hexane-AcOEt $92: 8$ ) to give the disilylated ( $\pm$ )lactam 9 as a colorless oil ( $35 \mathrm{mg}, 95 \%$ ). $v_{\max } / \mathrm{cm}^{-1} 1720$ (NCO) [Found (CI): $\mathrm{M}^{+}+1,372.2752 . \mathrm{C}_{19} \mathrm{H}_{41} \mathrm{NO}_{2} \mathrm{Si}_{2}+\mathrm{H}$ requires $M+1,263.1521] ; \delta_{\mathrm{H}}(500 \mathrm{MHz}) 3.64\left(2 \mathrm{H}, \mathrm{m}, \mathrm{OCH}_{2}\right), 3.37(1 \mathrm{H}$, dt, $J 10.5,2.5,4-\mathrm{H}), 2.79(1 \mathrm{H}$, ddd, $J 7.5,5.5,2.5,3-\mathrm{H}), 2.15-$ $1.54\left(4 \mathrm{H}, \mathrm{m}, 4-\mathrm{CH}_{2}\right.$ and $\left.3-\mathrm{CH}_{2}\right), 1.01(3 \mathrm{H}, \mathrm{t}, J 7, \mathrm{Me}), 0.96$ and 0.86 (each $9 \mathrm{H}, \mathrm{s}, \mathrm{Bu}^{t} \times 2$ ), 0.24 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{SiMe}$ ), 0.21 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{SiMe}$ ) and $0.04\left(6 \mathrm{H}, \mathrm{s}, \mathrm{SiMe}_{2}\right)$. The ${ }^{1} \mathrm{H}$ NMR and IR spectra were identical with those of an authentic sample. ${ }^{13,19}$

## ( $E, S$ )-4-(1-Methylethyl)-3-[2-ethyl-1-oxo-5-(benzyloxy)pent-2-enyl]oxazolidin-2-one 33

A solution of the ester $24(553 \mathrm{mg}, 2.1 \mathrm{mmol})$ and $\mathrm{LiOH}(170$ $\mathrm{mg}, 7 \mathrm{mmol})$ in $\mathrm{THF}-\mathrm{H}_{2} \mathrm{O}-\mathrm{MeOH}(1: 1: 2)\left(4 \mathrm{~cm}^{3}\right)$ was stirred at room temperature for 3.5 h . The mixture was acidified with $10 \%$ hydrochloric acid at $0{ }^{\circ} \mathrm{C}$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was washed, dried and concentrated to give the residue which was purified by $\mathrm{SCC}\left(\mathrm{AcOEt}-\mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ 1:1) and recrystallized from $n$-hexane to give the corresponding carboxylic acid 31a ( $335 \mathrm{mg}, 68 \%$ ) as colorless crystals $\mathrm{mp} 47-$ $48^{\circ} \mathrm{C} . v_{\text {max }} / \mathrm{cm}^{-1} 3100-2500$ and $1686(\mathrm{COOH})$ (Found: C, $71.77 ; \mathrm{H}, 7.74 . \mathrm{C}_{4} \mathrm{H}_{18} \mathrm{O}_{3}$ requires C, $\left.71.49 ; \mathrm{H}, 7.76 \%\right) ; \delta_{\mathrm{H}}(200$ $\mathrm{MHz}) 10.71(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 7.24(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 6.50(1 \mathrm{H}, \mathrm{t}, J 7$, $3-\mathrm{H}), 4.48\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{Ph}\right), 3.53\left(2 \mathrm{H}, \mathrm{t}, J 7,5-\mathrm{H}_{2}\right), 2.50(2 \mathrm{H}, \mathrm{q}$, $\left.J 7,4-\mathrm{H}_{2}\right), 2.26\left(2 \mathrm{H}, \mathrm{q}, J 7,2-\mathrm{CH}_{2}\right)$ and $1.01(3 \mathrm{H}, \mathrm{t}, J 7, \mathrm{Me})$.

A solution of $\mathrm{SOCl}_{2}\left(0.5 \mathrm{~cm}^{3}, 6.8 \mathrm{mmol}\right)$ and 31a $(110 \mathrm{mg}$,
$0.47 \mathrm{mmol})$ in benzene $\left(16 \mathrm{~cm}^{3}\right)$ was stirred under reflux for 1 h The solvent was removed under reduced pressure to give the acid chloride 31b ( 110 mg ) as a pale yellow oil. According to the literature, ${ }^{20}(S)$-4-(1-methylethyl)oxazolidin-2-one ( $58 \mathrm{mg}, 0.45$ mmol ) was lithiated with $n$-butyllithium in THF at $-78^{\circ} \mathrm{C}$ and successively acylated with the acid chloride 31b $(110 \mathrm{mg}, 0.47$ $\mathrm{mmol})$. The crude product was purified by MPCC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to give the imide $33(140 \mathrm{mg}, 86 \%)$ as a pale yellow oil; $[a]_{\mathrm{D}}^{28}+33.7$ (c $3.09, \mathrm{CHCl}_{3}$ ); $v_{\text {max }} / \mathrm{cm}^{-1} 1778$ and 1682 (CONHCOO) (Found: $\mathrm{M}^{+}, 345.1942 . \mathrm{C}_{20} \mathrm{H}_{27} \mathrm{NO}_{4}$ requires $M, 345.1939$ ); $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.25(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 6.00\left(1 \mathrm{H}, \mathrm{t}, J 6,3^{\prime}-\mathrm{H}\right), 4.55(2 \mathrm{H}$, $\left.\mathrm{s}, \mathrm{OCH}_{2} \mathrm{Ph}\right), 4.54(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 4.32(1 \mathrm{H}, \mathrm{t}, J 8,5-\mathrm{H}), 4.19(1 \mathrm{H}$, dd, $J 8,6,5-\mathrm{H}), 3.56\left(2 \mathrm{H}, \mathrm{t}, J 7,5^{\prime}-\mathrm{H}_{2}\right), 2.60-2.24(5 \mathrm{H}, \mathrm{m}$, $4^{\prime}-\mathrm{H}_{2}, 2^{\prime}-\mathrm{CH}_{2}$ and $\left.4-\mathrm{CH}\right), 1.03\left(3 \mathrm{H}, \mathrm{t}, J 8, \mathrm{CH}_{2} \mathrm{Me}\right)$ and 0.90 ( $6 \mathrm{H}, \mathrm{d}, \mathrm{J} 8, \mathrm{CHMe} \mathrm{C}_{2}$ ).

Addition of benzenethiol to chiral imide 33 (Table 1, entry 3)
Benzenethiol ( $1 \mathrm{~cm}^{3}, 10.8 \mathrm{mmol}$ ) was added with stirring at $0{ }^{\circ} \mathrm{C}$ to a solution of $n$-butyllithium ( 1.63 m solution in hexane; 2.3 $\left.\mathrm{cm}^{3}, 3.6 \mathrm{mmol}\right)$ in THF $\left(5 \mathrm{~cm}^{3}\right)$ to give a solution of a $2: 1$ mixture of benzenethiol and lithium benzenethiolate. To the resulting solution was added a solution of $33(250 \mathrm{mg}, 0.72$ $\mathrm{mmol})$ in THF $\left(5 \mathrm{~cm}^{3}\right)$ at $-78^{\circ} \mathrm{C}$. After being stirred at $-78^{\circ} \mathrm{C}$ for 5 h , the mixture was made alkaline by adding $5 \%$ aqueous NaOH and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The extract was dried and concentrated to give the residue which was purified by MPCC ( $n$-hexane-AcOEt $3: 1$ ) to give ( $2^{\prime} S, 3^{\prime} S, 4 S$ )-4-(1-methylethyl)-3-[2'-ethyl-1'-oxo-5'-benzyloxy-3'-(phenylsulfanyl)pentyl]oxa-zolidin-2-one 34 as a colorless oil ( $257 \mathrm{mg}, 78 \%$ ) and ( $2^{\prime} R, 3^{\prime} R, 4 S$ )-4-(1-methylethyl)-3-[2'-ethyl-1'-oxo-5'-benzyl-oxy- $3^{\prime}$-(phenylsulfanyl)pentyl]oxazolidin-2-one 35 as a colorless oil ( $64 \mathrm{mg}, 19 \%$ ). 34: $[a]_{\mathrm{D}}^{26}-2.5$ (c 2.77, $\mathrm{CHCl}_{3}$ ); $v_{\text {max }} / \mathrm{cm}^{-1}$ 1778 and 1692 (CONCOO) (Found: $\mathrm{M}^{+}, 455.2110 . \mathrm{C}_{26} \mathrm{H}_{33^{-}}$ $\mathrm{NO}_{4} \mathrm{~S}$ requires $M$, 455.2128) (Found: C, 67.7; H, 7.3; N, 3.0. $\mathrm{C}_{26} \mathrm{H}_{33} \mathrm{NO}_{4} \mathrm{~S} \cdot \frac{1}{4} \mathrm{H}_{2} \mathrm{O}$ requires C, 67.9; H, 7.3; N, 3.0\%); $\delta_{\mathrm{H}}(500$ $\mathrm{MHz}) 7.39-7.21(10 \mathrm{H}, \mathrm{m}, \mathrm{Ph} \times 2), 4.49(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 4.48$ and $4.45\left(2 \mathrm{H}, \mathrm{ABq}, J 12, \mathrm{OCH}_{2} \mathrm{Ph}\right), 4.24(1 \mathrm{H}, \mathrm{t}, J 9,5-\mathrm{H}), 4.18(2 \mathrm{H}$, $\mathrm{m}, 3^{\prime}-$ and $\left.5-\mathrm{H}\right), 3.79\left(1 \mathrm{H}, \mathrm{dt}, J 10,5.5,5^{\prime}-\mathrm{H}\right), 3.70(1 \mathrm{H}$, ddd, $\left.J 10,7,5,5^{\prime}-\mathrm{H}\right), 3.61(1 \mathrm{H}$, ddd, $J 10,6,4,2-\mathrm{H}), 2.27(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{C} H \mathrm{Me}_{2}\right), 2.13\left(1 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}\right), 1.95-1.78\left(3 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}\right.$ and $\left.\mathrm{CH}_{2} \mathrm{Me}\right), 0.89(3 \mathrm{H}, \mathrm{d}, J 7, \mathrm{CH} M e), 0.87\left(3 \mathrm{H}, \mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{Me}\right)$ and $0.80(3 \mathrm{H}, \mathrm{d}, J 7, \mathrm{CH} M e) ; \delta_{\mathrm{C}} 174.3$ (s), 153.7 (s), 138.6 (s), 135.6 (s), 131.6 (d), 128.9 (d), 128.3 (d), 127.6 (d), 127.4 (d), 126.8 (d), 72.8 ( t , , 67.6 ( t$), 63.0$ (t), 58.6 (d), 48.9 (d), 48.6 (d), 33.2 (t), 28.4 (d), 22.6 (t), 18.1 (q), 14.6 (q), 11.6 (q). $35:[a]_{\mathrm{D}}^{26}$ $+77.2\left(c 2.63, \mathrm{CHCl}_{3}\right) ; v_{\max } / \mathrm{cm}^{-1} 1777$ and $1695(\mathrm{CONCOO})$ (Found: $\mathrm{M}^{+}, 455.2110 . \mathrm{C}_{26} \mathrm{H}_{33} \mathrm{NO}_{4} \mathrm{~S}$ requires $M, 455.2128$ ) (Found: C, 66.8; H, 7.4; N, 3.0. $\mathrm{C}_{26} \mathrm{H}_{33} \mathrm{NO}_{4} \mathrm{~S} \cdot \frac{2}{3} \mathrm{H}_{2} \mathrm{O}$ requires C, $66.8 ; \mathrm{H}, 7.2 ; \mathrm{N}, 3.0 \%) ; \delta_{\mathrm{H}}(500 \mathrm{MHz}) 7.40-7.18(10 \mathrm{H}, \mathrm{m}$, $\mathrm{Ph} \times 2), 4.50$ and $4.44\left(2 \mathrm{H}, \mathrm{ABq}, J 12, \mathrm{OCH}_{2} \mathrm{Ph}\right), 4.24(1 \mathrm{H}$, ddd, $J 9,4,3,4-\mathrm{H}), 4.15\left(1 \mathrm{H}\right.$, ddd, $\left.J 9,7.5,5,3^{\prime}-\mathrm{H}\right), 4.08(1 \mathrm{H}$ dd, $J 9,3,5-\mathrm{H}), 3.96(1 \mathrm{H}, \mathrm{t}, J 9,5-\mathrm{H}), 3.78-3.69\left(2 \mathrm{H}, \mathrm{m}, 5^{\prime}-\mathrm{H}_{2}\right)$, $3.59\left(1 \mathrm{H}\right.$, br ddd, $\left.J 10,7,3.5,2^{\prime}-\mathrm{H}\right), 2.33(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{CH}), 2.08-$ $1.96\left(2 \mathrm{H}, \mathrm{m}, 2^{\prime}-\mathrm{CH}_{2}\right), 1.92-1.82\left(2 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}_{2}\right), 0.91$ and 0.84 (each $3 \mathrm{H}, \mathrm{d}, J 7, \mathrm{CHMe} e_{2}$ ) and $0.89\left(3 \mathrm{H}, \mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{Me}\right) ; \delta_{\mathrm{C}}$ 174.3 (s), 153.6 (s), 138.5 (s), 135.7 (s), 131.9 (d), 128.9 (d), 128.3 (d), 127.6 (d), 127.5 (d), 126.9 (d), 72.9 (t), 67.6 (t), $63.0(\mathrm{t}), 58.7$ (d), 48.6 (d), 48.2 (d), 32.7 (t), 28.5 (d), 23.1 (t), 18.0 (q), 14.6 (q), 10.9 (q).

## Conversion of ( $\mathbf{2}^{\prime} S, \mathbf{3}^{\prime} \boldsymbol{S}$ )-34 into ( $\mathbf{2 S , 3 S}$ )-28

Trimethylaluminium ( 2 m solution in hexane; $0.23 \mathrm{~cm}^{3}, 0.45$ mmol ) was added with stirring at $0^{\circ} \mathrm{C}$ to a suspension of methoxylamine hydrochloride ( $38 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) in THF ( 0.3 $\mathrm{cm}^{3}$ ). The resulting solution was added at $0^{\circ} \mathrm{C}$ to a solution of $\left(2^{\prime} S, 3^{\prime} S\right)$ - $34(25 \mathrm{mg}, 0.05 \mathrm{mmol})$ in THF $\left(5 \mathrm{~cm}^{3}\right)$. After being stirred at $0^{\circ} \mathrm{C}$ for 7 h , the mixture was acidified with $10 \%$ hydrochloric acid and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was washed, dried and concentrated to give the residue which was purified by MPCC $\left(\mathrm{AcOEt}-\mathrm{CH}_{2} \mathrm{Cl}_{2} 5: 1\right)$ to give $(2 S, 3 S)$ -

2-ethyl- $N$-methoxy-5-(benzyloxy)-3-(phenylsulfanyl)pentanamide $28(1 \mathrm{mg}, 5 \%)$ as a colorless oil and ( $1^{\prime} S, 2 S, 3 S$ )-2-ethyl-$N$-[2'-(methoxyaminocarbonyloxy)-1'-(1-methylethyl)ethyl]-5-benzyloxy-3-(phenylsulfanyl)pentanamide 38 ( $21 \mathrm{mg}, 84 \%$ ) as a colorless oil. The ${ }^{1} \mathrm{H}$ NMR and IR spectra of $(2 S, 3 S)-28$ were identical with those of $( \pm)-\mathbf{2 8}$. 28: $[a]_{\mathrm{D}}^{26}+3.4\left(c 2.54, \mathrm{CHCl}_{3}\right)$.
38: $v_{\max } / \mathrm{cm}^{-1} 3436(\mathrm{NH}), 1742,1670$ (NCO) (Found: $\mathrm{M}^{+}$, 502.2491. $\mathrm{C}_{27} \mathrm{H}_{38} \mathrm{~N}_{2} \mathrm{O}_{5} \mathrm{~S}$ requires $\left.M, 502.2499\right)$; $\delta_{\mathrm{H}}(500 \mathrm{MHz})$ $8.30(1 \mathrm{H}, \mathrm{br}$ s, CONHOMe), 7.38-7.20 ( $10 \mathrm{H}, \mathrm{m}, \mathrm{Ph} \times 2$ ), 5.64 $(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J 9, \mathrm{CONH}), 4.50$ and $4.47(2 \mathrm{H}, \mathrm{ABq}, J 12$, $\left.\mathrm{OCH}_{2} \mathrm{Ph}\right), 4.29\left(1 \mathrm{H}, \mathrm{dd}, J 11,3.5,2^{\prime}-\mathrm{H}\right), 4.12\left(1 \mathrm{H}, \mathrm{m}, 1^{\prime}-\mathrm{H}\right)$, $3.92\left(1 \mathrm{H}, \mathrm{dd}, J 11,9,2^{\prime}-\mathrm{H}\right), 3.83$ (1H, ddd, $\left.J 9,7,6,5-\mathrm{H}\right), 3.64$ $(1 \mathrm{H}, \mathrm{dt}, J 9,7,5-\mathrm{H}), 3.35(1 \mathrm{H}, \mathrm{td}, J 9,3.5,3-\mathrm{H}), 2.13-1.96(3 \mathrm{H}$, $\mathrm{m}, 2-\mathrm{CH}_{2}$ and $4-\mathrm{H}$ ), $1.76-1.65\left(3 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}, 4-\mathrm{H}\right.$ and $\left.1^{\prime}-\mathrm{CH}\right)$, 0.92 and 0.85 (each $3 \mathrm{H}, \mathrm{d}, J 7, \mathrm{CH} M e_{2}$ ) and $0.88(3 \mathrm{H}, \mathrm{t}, J 7.5$, $\left.\mathrm{CH}_{2} \mathrm{Me}\right) ;[a]_{\mathrm{D}}^{24}-5.3\left(c 0.94, \mathrm{CHCl}_{3}\right)$.

## (2R,3S)-S-Benzyl 2-ethyl-5-benzyloxy-3-(phenylsulfanyl)pentanethioate 36

(A) Using aluminium thiobenzyloxy 'ate' complex. Phenylmethanethiol ( $0.08 \mathrm{~cm}^{3}, 0.7 \mathrm{mmol}$ ) and trimethylaluminium ( 2 m solution in hexane; $0.45 \mathrm{~cm}^{3}, 0.9 \mathrm{mmol}$ ) were added dropwise with stirring at $0{ }^{\circ} \mathrm{C}$ to a solution of $n$-butyllithium ( 1.63 m solution in hexane; $\left.0.45 \mathrm{~cm}^{3}, 0.7 \mathrm{mmol}\right)$ in THF ( $6 \mathrm{~cm}^{3}$ ). After being stirred at $0{ }^{\circ} \mathrm{C}$ for 0.5 h , the resulting solution was added at $0^{\circ} \mathrm{C}$ to a solution of $\left(2^{\prime} S, 3^{\prime} S\right)-34(159 \mathrm{mg}, 0.35 \mathrm{mmol})$ in THF $\left(10 \mathrm{~cm}^{3}\right)$. After being stirred at $0{ }^{\circ} \mathrm{C}$ for 1.5 h , the mixture was acidified with $10 \%$ hydrochloric acid and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was washed with $5 \%$ aqueous NaOH , dried and concentrated to give the residue which was purified by MPCC ( $n$-hexane- $\mathrm{CH}_{2} \mathrm{Cl}_{2} 4: 1$ ) to give $(2 R, 3 S)$-36 ( $153 \mathrm{mg}, 97 \%$ ) as a colorless oil; $v_{\text {max }} / \mathrm{cm}^{-1} 1676$ (COS) (Found: $\mathrm{M}^{+}$, 450.1701. $\mathrm{C}_{27} \mathrm{H}_{30} \mathrm{O}_{2} \mathrm{~S}_{2}$ requires $M$, 450.1686); $\delta_{\mathrm{H}}(500 \mathrm{MHz}) 7.27(15 \mathrm{H}, \mathrm{m}, \mathrm{Ph} \times 3), 4.44$ and $4.41(2 \mathrm{H}, \mathrm{ABq}$, $\left.J 12, \mathrm{OCH}_{2} \mathrm{Ph}\right), 4.15$ and $4.11\left(2 \mathrm{H}, \mathrm{ABq}, J 14, \mathrm{SCH}_{2} \mathrm{Ph}\right), 3.75$ (1H, td, $J 9,5,5-\mathrm{H}), 3.65(1 \mathrm{H}, \mathrm{ddd}, J 9,6,4,5-\mathrm{H}), 3.50(1 \mathrm{H}$, ddd, $J 10,6.5,5,3-\mathrm{H}), 2.70(1 \mathrm{H}$, ddd, $J 10,7,5,2-\mathrm{H}), 2.06-2.13$ $(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 1.79-1.90\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{CH}_{2}\right), 1.72(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H})$ and $0.86(3 \mathrm{H}, \mathrm{t}, J 7.5, \mathrm{Me}) ;[a]_{\mathrm{D}}^{27}-71.1\left(c 2.90, \mathrm{CHCl}_{3}\right)$.
(B) Using lithium benzylthiolate. Phenylmethanethiol ( 0.028 $\mathrm{cm}^{3}, 0.24 \mathrm{mmol}$ ) was added dropwise with stirring at $0{ }^{\circ} \mathrm{C}$ to a solution of $n$-butyllithium ( 1.63 m solution in hexane; $0.1 \mathrm{~cm}^{3}$, $0.18 \mathrm{mmol})$ in THF $\left(3 \mathrm{~cm}^{3}\right)$. After being stirred at $0^{\circ} \mathrm{C}$ for 0.5 h , the resulting solution was added at $0^{\circ} \mathrm{C}$ to a solution of ( $2^{\prime} S, 3^{\prime} S$ ) $\mathbf{3 4}\left(56 \mathrm{mg}, 0.12 \mathrm{mmol}\right.$ ) in THF ( $3 \mathrm{~cm}^{3}$ ). After being stirred at $0^{\circ} \mathrm{C}$ for 1.5 h , the mixture was made alkaline by addition of $5 \%$ aqueous NaOH and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was washed, dried and concentrated to give the residue which was purified by MPCC ( $n$-hexane- $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $4: 1)$ to give $(2 R, 3 S)-36(34 \mathrm{mg}, 63 \%)$ as a colorless oil. The ${ }^{1} \mathrm{H}$ NMR spectra and IR spectra of $(2 R, 3 S)-\mathbf{3 6}$ were identical with those of the sample prepared in (A).

## (2S,3R)-S-Benzyl 2-ethyl-5-benzyloxy-3-(phenylsulfanyl)pentanethioate 37

According to the procedure described for preparation of $(2 R, 3 S)-\mathbf{3 6}$, a solution of $\left(2^{\prime} R, 3^{\prime} R\right)-\mathbf{3 5}(75 \mathrm{mg}, 0.16 \mathrm{mmol})$ and aluminium thiobenzyloxy 'ate' complex in THF was stirred at $0{ }^{\circ} \mathrm{C}$ for 1.5 h . The crude product was purified by MPCC ( $n$-hexane- $\mathrm{CH}_{2} \mathrm{Cl}_{2} 4: 1$ ) to give ( $2 S, 3 R$ )-37 ( $68 \mathrm{mg}, 95 \%$ ). The ${ }^{1} \mathrm{H}$ NMR and IR spectra of $\mathbf{3 7}$ were identical with those of $\mathbf{3 6}$. $[a]_{\mathrm{D}}^{27}+70.3\left(c 2.56, \mathrm{CHCl}_{3}\right)$.

## (2S,3S)-2-Ethyl-5-benzyloxy-3-(phenylsulfanyl)pentanoic acid 39

(A) Using silver trifluoroacetate. Silver trifluoroacetate (260 $\mathrm{mg}, 1.2 \mathrm{mmol}$ ) was added with stirring to a solution of the thioester $36(41 \mathrm{mg}, 0.09 \mathrm{mmol})$ in THF- $\mathrm{H}_{2} \mathrm{O}(3: 1)\left(2 \mathrm{~cm}^{3}\right)$. After being stirred under reflux for 5 h , the mixture was acidified with $10 \%$ hydrochloric acid and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$.

The organic layer was washed, dried and concentrated to give the residue which was purified by $\mathrm{SCC}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH} 10: 1\right)$ to give the carboxylic acid 39 ( $30 \mathrm{mg}, 96 \%$ ) as a colorless amorphous solid; $v_{\text {max }} / \mathrm{cm}^{-1} 3000-2600,1692(\mathrm{COOH})$ (Found: $\mathrm{M}^{+}$, 344.1449. $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{O}_{3} \mathrm{~S}$ requires $\left.\mathrm{M}, 344.1445\right)$; $\delta_{\mathrm{H}}(500 \mathrm{MHz})$ $7.50-7.22(10 \mathrm{H}, \mathrm{m}, \mathrm{Ph} \times 2), 4.50\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{Ph}\right), 3.92-3.60$ $\left(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{CH}_{2}\right), 3.60-3.46(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 2.60-2.48(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H})$, $2.20-1.64\left(4 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}_{2}\right.$ and $\left.2-\mathrm{CH}_{2}\right)$ and $0.94(3 \mathrm{H}, \mathrm{t}, J 8, \mathrm{Me})$.
(B) Using silver perchlorate. According to method A, a solution of the thioester $\mathbf{3 6}(20 \mathrm{mg}, 0.04 \mathrm{mmol})$ and silver perchlorate ( $130 \mathrm{mg}, 0.63 \mathrm{mmol}$ ) in THF- $\mathrm{H}_{2} \mathrm{O}(3: 1)\left(0.8 \mathrm{~cm}^{3}\right)$ was stirred under reflux for 5 h . The crude product was purified by SCC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH} 10: 1\right)$ to give the carboxylic acid 39 (6 $\mathrm{mg}, 45 \%$ ) as a colorless amorphous solid. The ${ }^{1} \mathrm{H}$ NMR and IR spectra of 39 were identical with those of the sample prepared in (A).

## (2S,3S)-2-Ethyl- $N$-methoxy-5-benzyloxy-3-(phenylsulfanyl)-

 pentanamide 28A solution of methoxylamine hydrochloride ( $44 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) in $\mathrm{H}_{2} \mathrm{O}\left(0.25 \mathrm{~cm}^{3}\right)$ was added under stirring at room temperature to a solution of the carboxylic acid $39(90 \mathrm{mg}, 0.26 \mathrm{mmol})$ in THF- $\mathrm{H}_{2} \mathrm{O}(6: 1)\left(0.5 \mathrm{~cm}^{3}\right)$. Then a solution of 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (WSC) ( $96 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) in $\mathrm{H}_{2} \mathrm{O}\left(0.25 \mathrm{~cm}^{3}\right)$ was added under stirring at room temperature to the resulting mixture. After being stirred for 5 h , the mixture was acidified with $10 \%$ hydrochloric acid and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was washed, dried and concentrated to give the residue which was purified by SCC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH} 10: 1\right)$ to give the $(2 S, 3 S)$ hydroxamate $\mathbf{2 8}(83 \mathrm{mg}, 86 \%) . \mathbf{2 8}$ was identical with the product 28 formed from 34 based on comparison of their spectra and optical rotation.

## (3R, trans)-3-Ethyl-1-methoxy-4-[2-(benzyloxy)ethyl]azetidin-2one 29

According to the procedure described for the preparation of the trans-13, the $(2 S, 3 S)-\mathbf{2 8}(112 \mathrm{mg}, 0.3 \mathrm{mmol})$ was alkylated with methyl iodide in the presence of silver perchlorate and then the resulting sulfonium salt was treated with $\mathrm{K}_{2} \mathrm{CO}_{3}$. The crude product was purified by MPCC $\left(\mathrm{AcOEt}-\mathrm{CH}_{2} \mathrm{Cl}_{2} 1: 10\right)$ to give ( $3 R, 4 R$ )-29 (trans: cis $=97: 3$ ) as a colorless oil ( $65 \mathrm{mg}, 83 \%$ ). The ratio of trans- to cis-products was determined by 200 MHz ${ }^{1} \mathrm{H}$ NMR spectroscopy. The ${ }^{1} \mathrm{H}$ NMR and IR spectra of $(3 R, 4 R)-\mathbf{2 9}$ were identical with those of $( \pm)-\mathbf{2 9} ;[a]_{\mathrm{D}}^{26}+1.32$ (c $2.27, \mathrm{CHCl}_{3}$ ).

## (3R, trans)-3-Ethyl-1-[(1,1-dimethylethyl)dimethylsilyl]-4-\{2-[(1,1-dimethylethyl)dimethylsilyloxy]ethyl\}azetidin-2-one 9

According to the procedure described for the preparation of $( \pm)-\mathbf{9},(3 R, 4 R)-\mathbf{2 9}$ (trans:cis $=97: 3)(59 \mathrm{mg}, 0.2 \mathrm{mmol})$ was converted into the disilylated ( $3 R$, trans) - $\beta$-lactam $9(70 \mathrm{mg}$, $91 \%$ ) as a colorless oil. The ${ }^{1} \mathrm{H}$ NMR and IR spectra of ( $3 R$, trans) $-\beta$-lactam 9 were identical with those of $( \pm)-9 ;[a]_{\mathrm{D}}^{26}$ $-34.2\left(c 0.79, \mathrm{CHCl}_{3}\right)\left\{\right.$ lit. $[a]_{\mathrm{D}}-37.73\left(c 2.25, \mathrm{CHCl}_{3}\right){ }^{19}-30.1$ $\left.\left(c 2.9, \mathrm{CHCl}_{3}\right)^{13}\right\}$.

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